

Astrophysics I: Stars and Stellar Evolution

AST 4001

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Stars and Stellar Evolution, Fall 2008

Overview

- 1 Recap
 - Local Stability
 - Convection
- 2 Advanced Nuclear Burning Stages
 - Overview – Origin of the Elements
 - Helium Burning and Beyond

Summary on Local Stability

- Convection according to **Ledoux Criterion** when

$$\nabla_{\text{rad}} > \nabla_{\text{ad}} + \frac{\varphi}{\delta} \nabla_{\mu}$$

- Convection according to **Schwarzschild Criterion** when

$$\nabla_{\text{rad}} > \nabla_{\text{ad}}$$

- **Semiconvection** when

$$\frac{\varphi}{\delta} \nabla_{\mu} > 0, \nabla_{\text{rad}} < \nabla_{\text{ad}} + \frac{\varphi}{\delta} \nabla_{\mu}$$

- **Thermohaline convection** when

$$\frac{\varphi}{\delta} \nabla_{\mu} < 0, \nabla_{\text{rad}} < \nabla_{\text{ad}} + \frac{\varphi}{\delta} \nabla_{\mu}$$

Summary on Convection

- In the stellar interior bubbles rise close to adiabatically, and the temperature gradient in the convection zone is close to adiabatic, but slightly steeper

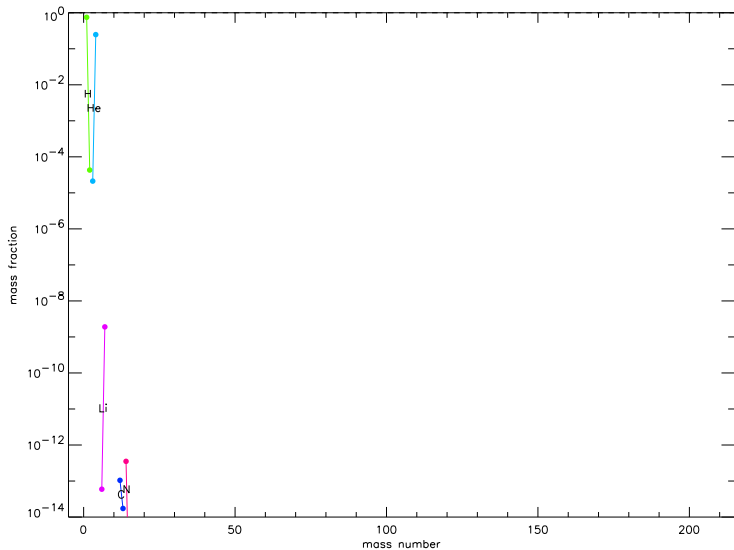
$$\frac{dS}{dr} \lesssim 0$$

- The four temperature gradients in convection zone are in order of increasing steepness
 - adiabatic temperature gradient
 - temperature gradient of rising bubble (i.e., “up-flow”)
 - temperature gradient of surrounding media
 - (factious) *radiative* temperature gradient
- Convection zones are “well mixed” – close to chemically homogeneous

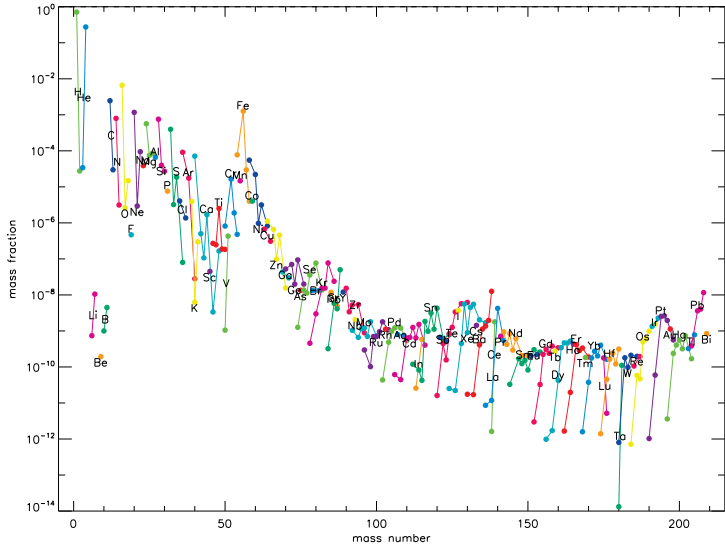
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Composition of the Universe after the Big Bang



The Composition of the Sun

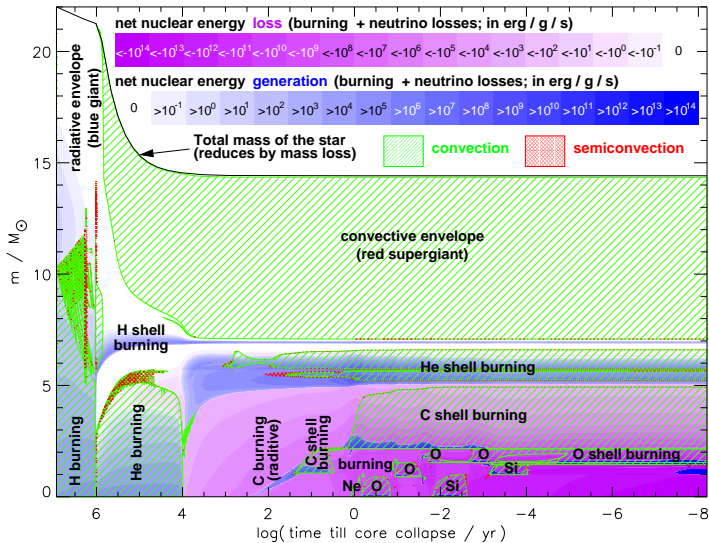


Overview - Burning Phases in Stars

20 M_⊙ star

Fuel	Main Product	Secondary Product	T (10 ⁹ K)	Time (yr)	Main Reaction
H	He	¹⁴ N	0.02	10 ⁷	^{CNO} 4 H → ⁴ He
He	O, C	¹⁸ O, ²² Ne s-process	0.2	10 ⁶	3 He ⁴ → ¹² C ¹² C(α,γ) ¹⁶ O
C	Ne, Mg	Na	0.8	10 ³	¹² C + ¹² C
Ne	O, Mg	Al, P	1.5	3	²⁰ Ne(γ,α) ¹⁶ O ²⁰ Ne(α,γ) ²⁴ Mg
O	Si, S	Cl, Ar, K, Ca	2.0	0.8	¹⁶ O + ¹⁶ O
Si, S	Fe	Ti, V, Cr, Mn, Co, Ni	3.5	0.02	²⁸ Si(γ,α)...

Overview - Burning Phases in the Stellar Interior



Stellar Structure Equations - Nuclear Burning

stationary terms

time-dependent terms

$$\frac{\partial r}{\partial m} = \frac{1}{4\pi r^2 \rho} \quad (1)$$

$$\frac{\partial P}{\partial m} = -\frac{Gm}{4\pi r^4} - \frac{1}{4\pi r^2} \frac{\partial^2 r}{\partial t^2} \quad (2)$$

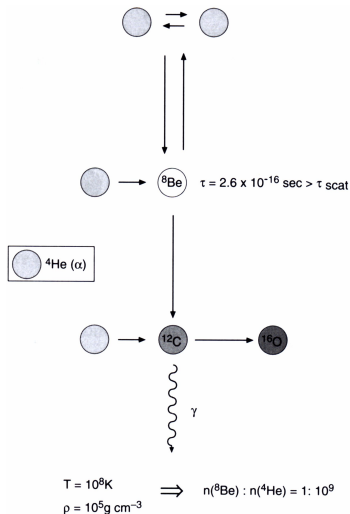
$$\frac{\partial F}{\partial m} = \epsilon_{\text{nuc}} - \epsilon_{\nu} - c_P \frac{\partial T}{\partial t} + \frac{\delta}{\rho} \frac{\partial P}{\partial t} \quad (3)$$

$$\frac{\partial T}{\partial m} = -\frac{GmT}{4\pi r^4 P} \nabla \left[1 + \frac{r^2}{Gm} \frac{\partial^2 r}{\partial t^2} \right] \quad (4)$$

$$\frac{\partial X_i}{\partial t} = f_i(\rho, T, \mathbf{X}) \quad (5)$$

where $\mathbf{X} = \{X_1, X_2, \dots, X_i, \dots\}$.

The Triple- α Reaction

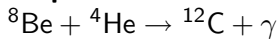


Step 1:



Built up equilibrium abundance of ${}^8\text{Be}$
 Lifetime of ${}^8\text{Be}$ is only $2.6 \times 10^{-16} \text{ s!}$

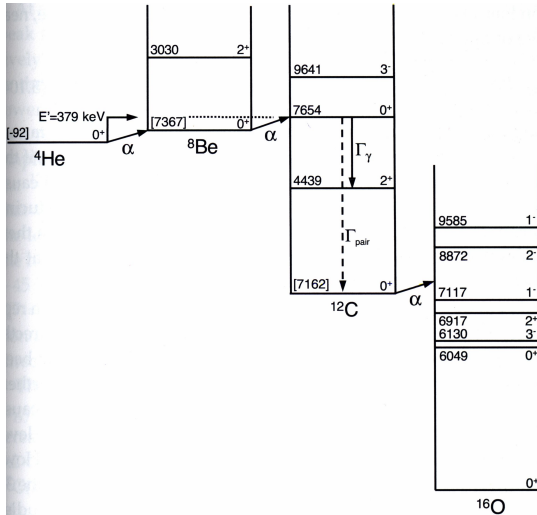
Step 2:



$$Q_{3\alpha} = 7.275 \text{ MeV}$$

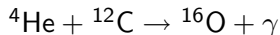
$$\langle \sigma v \rangle \propto \rho^2 T^{40}$$

Helium Burning level scheme



Additional Reactions of Helium Burning

Oxygen Production



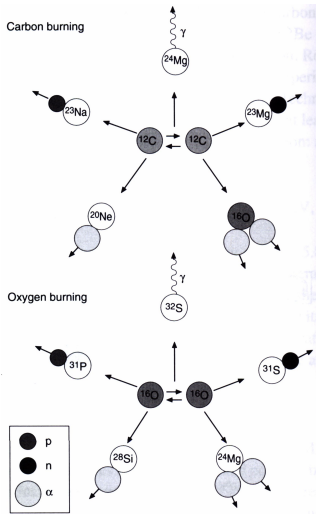
$$Q = 7.162 \text{ MeV}$$

$$\langle \sigma v \rangle \propto \rho T^{40}$$

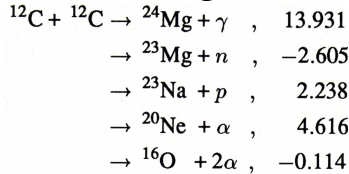
The final abundance of carbon is set by the competition of 3α and ${}^{12}\text{C}(\alpha, \gamma){}^{16}\text{O}$ reactions;

The production of ${}^{16}\text{O}$ can only start when a sufficient amount of ${}^{12}\text{C}$ has been made.

Carbon and Oxygen Burning

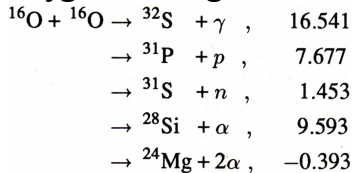


Carbon Burning



Average $Q = 13 \text{ MeV}$

Oxygen Burning



Average $Q = 16 \text{ MeV}$

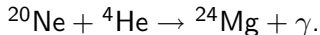
Neon Burning

Neon burning proceeds by a combination of photo-disintegrations and α captures:

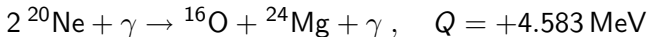


This reaction dominates over the inverse reaction known from helium burning for $T > 1.5 \times 10^9 \text{ K}$.

Subsequently, the ^4He is captured on another ^{20}Ne nucleus:



The net result is

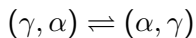


“Silicon” Burning

Actually, often we have more sulfur in the star than there is silicon, but it is custom to call this phase “silicon burning”.

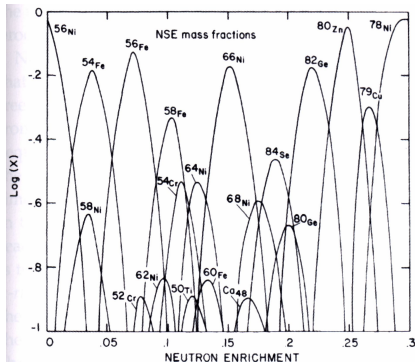
Typical burning temperature is $3 \dots 3.5 \times 10^9$ K.

Similar to neon burning, silicon burning proceeds as a series of photo-disintegration reactions, mostly, (γ, α) , and helium capture reactions, (α, γ) to build up iron group elements.



At the high T and ρ of these conditions, also *weak reactions* occur, converting protons into neutrons and leading to a *neutron excess*. This allows to actually make stable iron isotopes.

Beyond Silicon/Sulfur Burning



NSE distribution for
 $T = 3.5 \times 10^9 \text{ K}$,
 $\rho = 10^7 \text{ g/cm}^3$

After silicon burning T and ρ is so high that the nuclei are in **nuclear statistical equilibrium**, i.e., the reactions are fast compared to the evolution time-scale of the star, and the abundance distribution of the nuclei is given by a *Saha equation*.

Summary of Energetics

<i>Nuclear Fuel</i>	<i>Process</i>	$T_{threshold}$ $10^6 K$	<i>Products</i>	<i>Energy per Nucleon (MeV)</i>
H	$p-p$	~ 4	He	6.55
H	CNO	15	He	6.25
He	3α	100	C, O	0.61
C	$C + C$	600	O, Ne, Na, Mg	0.54
O	$O + O$	1000	Mg, S, P, Si	~ 0.3
Si	Nuc. eq.	3000	Co, Fe, Ni	< 0.18